



















WATER IMPURITIES COMPARISON CHART

Water Impurities	 Chlorine Chlorine (Cl) is fed into a water supply to control microorganisms or to oxidize contaminants like iron. Municipal treatment usually involves gas chlorination, while private supplies provide a bleach solution.	 Chloramine Municipal water treatment plants also disinfect water with chloramines (NH ₂ Cl), which is ammonia combined with chlorine treated water. Although not as effective at killing microorganisms as is free chlorine, chloramines prevent biogrowth and are more stable in the municipal distribution system.	 Chlorides Chloride in water is an anion (Cl ⁻) that is very soluble. It's most commonly associated with sodium, as in table salt, sodium chloride (NaCl). Common sources of chloride include seawater intrusion, road salt, and natural deposits in the earth.
How are they introduced?	 Municipal water treatment plants	 Municipal water treatment plants	 Naturally occurring in the ground
What problems do they cause?	 Taste, odor concerns, corrosion	 Taste, odor concerns, corrosion	 Corrosion
Where do you find them?	 Municipal water treatment plants	 Municipal water treatment plants	 Naturally occurring in the ground
How do you treat them?	 Standard carbon based filtration	 Specialized carbon filtration	 Reverse Osmosis specialized filtration
How do you detect them?	 On-site and/or Lab testing	 On-site and/or Lab testing	 On-site and/or Lab testing